

THE SYNTHESIS AND  
ANTIBACTERIAL ACTIVITIES OF  
NOVEL 7 $\alpha$  SUBSTITUTED  
CEPHALOSPORINS

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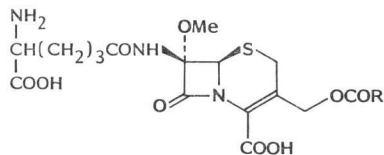
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The cephamycins are chemically related to the cephalosporin family of antibiotics<sup>1-3</sup>. A common feature of this group of naturally-occurring compounds exemplified by cephamycins A~C (**1**) is the presence of a 7 $\alpha$ -methoxy substituent which renders them highly resistant to hydrolysis by  $\beta$ -lactamases. Recent years have seen the preparation of many semisynthetic cephamycin derivatives, two of which, *viz.*, cefoxitin (**2a**)<sup>4</sup> and cefmetazole (**2b**)<sup>5</sup> are now used clinically. Certain semisynthetic penicillins bearing a 6 $\alpha$ -

methoxy substituent have been prepared in our laboratories, *e.g.* temocillin (**3**)<sup>6</sup>, and shown to possess significant antibacterial activity as well as stability towards  $\beta$ -lactamases. The synthesis of penicillins and cephalosporins bearing substituents other than methoxy at the 6 $\alpha$ (7 $\alpha$ ) position have also been reported but the compounds described were found to show reduced antibacterial activity<sup>7-10</sup> compared to their 6 $\alpha$ (7 $\alpha$ ) unsubstituted analogues. However, we have found that 6 $\alpha$ -hydroxymethyl analogues of certain penicillins (including piperacillin<sup>11</sup>), exemplified by (**4a**), possess antibacterial activity<sup>12</sup>. We have also prepared selected 7 $\alpha$ -hydroxymethylcephalosporins<sup>13</sup>, whose synthesis and *in vitro* antibacterial properties we now report.

Introduction of the hydroxymethyl (CH<sub>2</sub>OH) group into the 7 $\alpha$ -position of the cephalosporin nucleus involved procedures similar to those employed by the Merck group<sup>10</sup>. The ester (**5a**) was converted into the Schiff's base (**6a**) by treatment with *p*-nitrobenzaldehyde. Treatment of a *N,N*-dimethylformamide (DMF) solution of **6a** with anhydrous potassium carbonate formed the carbanion at the 7 $\alpha$ -position which was

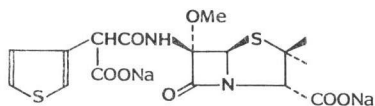


**1**

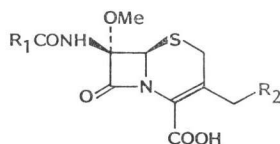
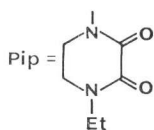
Cephamycin A R = *p*-HOSO<sub>2</sub>OC<sub>6</sub>H<sub>4</sub>CH=C(OMe)-

Cephamycin B R = *p*-HOC<sub>6</sub>H<sub>4</sub>CH=C(OMe)-

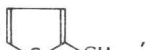
Cephamycin C R = NH<sub>2</sub>-

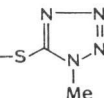


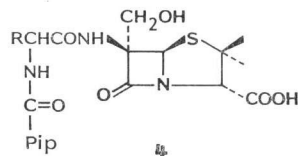
**3**




**2**


**a** R<sub>1</sub> = , R<sub>2</sub> = -OCONH<sub>2</sub>

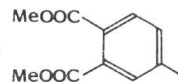
**b** R<sub>1</sub> = NCCH<sub>2</sub>SCH<sub>2</sub>-, R<sub>2</sub> = 



**4**

**a** R = 

**b** R = HO-

**c** R = 

quenched with gaseous formaldehyde giving the alcohol (**6b**). It was found that a temperature of  $-22^{\circ}\text{C}$  was critical in this reaction in order to avoid  $1^{\beta} \rightarrow 1^{\alpha}$  isomerisation whilst still maintaining an acceptable reaction rate.

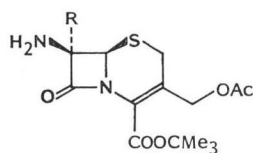
Condensation of formaldehyde with the  $7\beta$ -benzylideneaminocephem nucleus has been assigned by Merck workers as occurring at the  $7\alpha$ -position<sup>10</sup>. This assignment is strengthened in our case by the comparable activities of  $6\alpha$ - $\text{CH}_2\text{OH}$  piperacillin analogues (**4a~c**)<sup>12</sup> with the corresponding  $7\alpha$ - $\text{CH}_2\text{OH}$  cefoperazone analogues (**10a,c,d**) described below, and by the existence of a positive  $25 (\pm 5)\%$  nuclear Overhauser effect between the C- $6\alpha$  hydrogen and C-7  $\text{CH}_2$  group.

Conversion of **6b** into the amine (**5b**) was effected by treatment with preformed 2,4-dinitrophenylhydrazine tosylate salt in ethanol. The  $7\beta$ -acetamido derivatives (**7a,b,e**) were prepared by coupling the appropriate side-chain acids

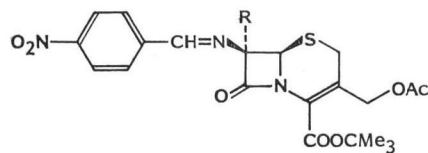
(**8a, b** and **9a**) to **5b** via their acid chlorides (**8c, d** and **9b**), generated using oxalyl chloride in the presence of a catalytic amount of DMF. At this point the protected *p*-hydroxyphenyl analogue (**7b**) was catalytically hydrogenated to the free hydroxyl compound **7c**. Successive treatment with trifluoroacetic acid and sodium ethyl hexanoate (SEH) in methyl isobutyl ketone (MIBK) transformed the esters (**7a, 7c** and **7e**) into the corresponding C-4 carboxylic acid sodium salts (**10a,c,e**) respectively.

The diacetoxyphenylcephalosporin ester (**7d**) was prepared by a *N,N*-dicyclohexylcarbodiimide mediated coupling of the appropriate side-chain acid (**11**) to the nucleus (**5b**). The ester (**7d**) was subsequently converted to the sodium salt (**10d**) in the usual way.

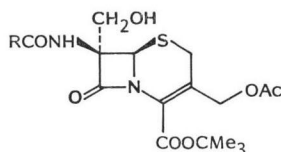
Certain  $7\alpha$ -hydroxymethyl substituted acylaminocephalosporins have been shown to possess significant antibacterial activity. It can be seen from Table 1 that compounds (**10a, c** and **d**)

**5**

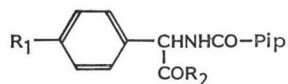
- a** R = H  
**b** R =  $\text{CH}_2\text{OH}$

**6**

- a** R = H  
**b** R =  $\text{CH}_2\text{OH}$

**7**

- a** R =  $\text{PhCHNHCO-Pip}$   
**b** R =  $\text{Z-O-C}_6\text{H}_4\text{-CHNHCO-Pip}$   
**c** R =  $\text{HO-C}_6\text{H}_4\text{-CHNHCO-Pip}$   
**d** R =  $\text{MeOOC-C}_6\text{H}_3(\text{MeOOC})\text{-CHNHCO-Pip}$   
**e** R =  $\text{S-C}_4\text{H}_3\text{-CHCOO-C}_6\text{H}_4\text{-Me}$

**8**

- a**  $\text{R}_1 = \text{H}, \text{R}_2 = \text{OH}$   
**b**  $\text{R}_1 = \text{O-Z}, \text{R}_2 = \text{OH}$   
**c**  $\text{R}_1 = \text{H}, \text{R}_2 = \text{Cl}$   
**d**  $\text{R}_1 = \text{O-Z}, \text{R}_2 = \text{Cl}$



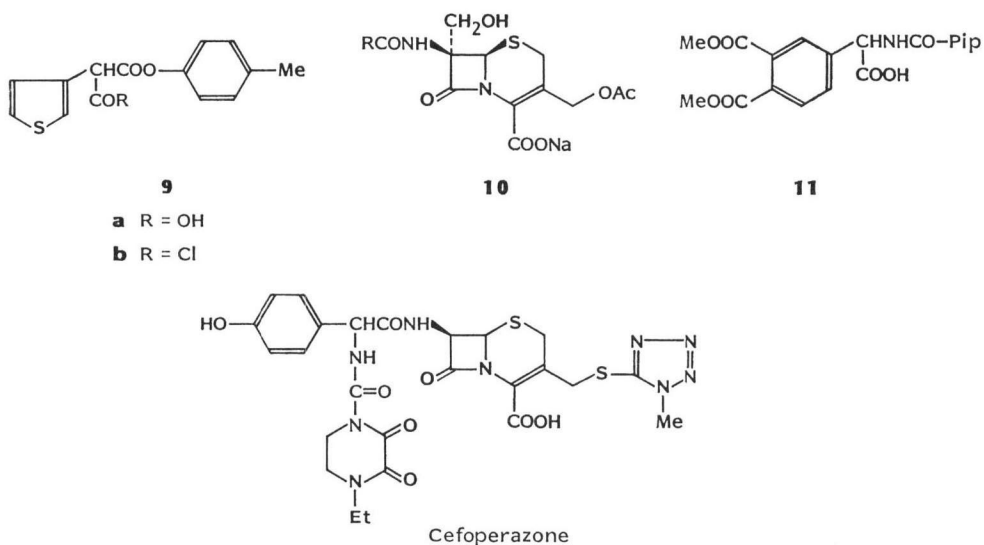
Table 1. The relative activities *in vitro* of 10a, 10c, 10d and cefoperazone.

Organism	MIC ( $\mu\text{g/ml}$ )*			
	10a	10c	10d	Cefoperazone
<i>Escherichia coli</i> ESS	<0.02	0.05	<0.06	<0.01
<i>E. coli</i> JT4 <sup>a</sup>	5.0	2.5	<0.06	>100
<i>E. coli</i> JT425 <sup>b</sup>	2.5	1.0	<0.06	1.0
<i>E. coli</i> NCTC 10418	2.5	1.0	<0.06	0.05
<i>Pseudomonas aeruginosa</i> NCTC 10662	>100	50	2.0	5.0
<i>P. aeruginosa</i> Dalglish <sup>a</sup>	>100	25	2.0	25
<i>Serratia marcescens</i> US 32	5.0	5.0	2.0	5.0
<i>Klebsiella pneumoniae</i> A	0.5	0.25	0.12	0.05
<i>Enterobacter cloacae</i> N1	10	10	8.0	1.0
<i>Proteus mirabilis</i> C977	2.5	2.5	16	0.5
<i>P. mirabilis</i> 889 <sup>b</sup>	1.0	0.5	2.0	10
<i>P. rettgeri</i>	10	10	8.0	2.5
<i>Staphylococcus aureus</i> Oxford	>100	>100	>128	1.0
<i>S. aureus</i> Russell <sup>a</sup>	>100	>100	>128	5.0
<i>Streptococcus pyogenes</i> CN10	100	>100	32	0.1

<sup>a</sup>  $\beta$ -Lactamase-producing strain (plasmid-mediated).

<sup>b</sup>  $\beta$ -Lactamase-producing strain (non-plasmid-mediated).

\* Determined by serial dilution in nutrient agar containing 5% defibrinated horse blood, inoculum 0.001 ml of an undiluted overnight broth culture (approximately  $10^8$  cfu).



which can be considered as structurally related to cefoperazone<sup>14)</sup>, inhibited many Gram-negative bacteria with MIC values of less than 10  $\mu\text{g/ml}$ . The less active of these three compounds, *viz.* **10a** and **10c**, still demonstrated improved activity compared with cefoperazone against certain  $\beta$ -lactamase producing organisms such as *Escherichia coli* JT4 and *Proteus mirabilis* 889, although they were usually less potent against other Gram-negative bacteria, including *E. coli* NCTC 10418

and *P. mirabilis* C977. Presumably the 7 $\alpha$ -hydroxymethyl group, like the 7 $\alpha$ -methoxy substituent in the cephamycin series, confers improved stability to  $\beta$ -lactamases.

However, the diacetoxyphenyl example (**10d**) showed much improved activity at least the equal of cefoperazone against the cefoperazone sensitive *E. coli* NCTC 10418 and *Pseudomonas aeruginosa* NCTC 10662, while again displaying stability to  $\beta$ -lactamases as expressed by its MIC values

against *E. coli* JT4, *P. aeruginosa* Dalglish and *P. mirabilis* C977.

All three compounds (10a, c and d) had little or no Gram-positive activity. The  $\alpha$ -carboxycephalosporin (10e) showed no significant antibacterial activity.

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